9/13/2022		COURCE							_/	/ISD	DIVISION	
	ELECTIVE	COURSE									NEW CO	
Ш	ELECTIVE	COURSE								V	KEVISION	V
				ke Land Colle	_							
				Course Information Fo	m							
COURSE NUMBER:	СН	M-101		TITLE: (30 Characters	Max)	Physic	al Science	e II				
SEM CR HRS:	3	Lecture:		2		Lab:	2				ECH:	4
	☑ Gen	Ed / IAI 🔲 C	areer/T	echnical	Clini	cal Practicum:	0		SOE/		SOE	0
Course Level:	☐ Bacc	calaureate /Non-IAI 🔲 D	ev Ed/	Not in Degree Audit	Clinic	cai Practicum:	0	Intern	ship:	0	ECH:	U
COURSE PCS #		11 - 40.0101		IAI Code		P1 9	903L		Contac	t Hours (M	inutes Per V	Veek)
Repeatable (Y/N):	N	Pass/Fail (Y/N):	N	Variable Credit (Y/N):	N	Min:	Max:	16	Wks	200	8 wks	400
Prerequisites:	Noi	ne										
Catalog Description: (40 W Limit)		introductory discussion of che uses of chemicals and applica				veen basic knov	vledge ne	eded to u	ndersta	and		
						Contact						
	List t	he Major Course Segments	(Units)			Lecture Hours	Contact Hour		Clini Practi		Non-C Internsh	
Careers in Chemistry						1	0					
Historical Review						1	0					
Metric System and Math Prin	ciples					3	2					
Atomic Theory						3	2					
Chemical Bonding						3	2					
Chemical Reactions						4	6					
Acid-Base Theory						4	6					
Organic Chemicals						4	6					
Drugs						2	2					
Oxidation and Reduction						2	2					
Chemicals and the Environm	ent				TOTAL	30	30		0		O	
					IOIAL	30	30		U		U	
				EVALUATION								
	IZZES 🔽	EXAM	clu	EVALUATION	OB	AL PRES 🗸				PAPERS		
IΔRV	VORK 🗸	PROJECT	<u> </u>			P FINAL 🗹				OTHER	片	
LAD	·OKKIE	FROJECT	기니		COM	· · · · · · · · · · · · · · · · · · ·				JIHER		
				COURSE MATERIALS								
	TITLE:	Chemistry for Changing Tim										
	AUTHOR:	John W. Hill										
PL	JBLISHER:	Pearson Publishing Compan	у									
VOLUME/EDIT	ION/URL:	12th										
COPYRIG	HT DATE:	2010		-								

MAJOR COURSE SEGMENT	HOURS	LEARNING OUTCOMES
		The student will be able to:
Careers in Chemistry	1	Describe career opportunities in chemistry and related areas.
Historical Review	1	Describe the contributions of scholars to structure and reactions of matter.
Metric System and Math Principles	5	Use the SI and English system for calculation of mass, volume, length, temperature, density and make conversions between the two systems.
Atomic Theory	5	Describe Dalton's atomic theory and distinguish between atomic weight and atomic number. Describe the organization of the periodic table and use the periodic table to predict trends in the elements.

Chemical Bonding		Draw name simple molecular and ionic compounds. Distinguish between the two main types of chemical bonds
Chemical Reactions		Balance chemical equations and predict the products when a chemical reaction occurs.
Acid-Base Theory	10	Define the terms acid, and base calculate the pH of solutions from hydronium or hydroxyl in concentration, perform acid-base titrations.
Organic Chemicals	10	Classify organic compounds according to functional groups. Present and give uses of several compounds in each class of compounds.
Drugs	4	List the major types of drugs used in medicine and the dangers of abuse of drugs.
Oxidation And Reduction	4	Identify the major keys to recognizing redox processes, utilize oxidation numbers, Understand the role of redox in nature
Chemicals and the Environment	5	List the major types and sources of air and water pollutants and methods of control of pollutants.
	60	

Demonstrate an understanding of basic chemical concepts & principles Be able to apply appropriate chemical concepts & principles, to the explanation of personal and societal experiences/problems Understand the general structure & properties of compounds found in society Evaluate scientific information found in the media.
Understand the general structure & properties of compounds found in society Evaluate scientific information found in the media.
Evaluate scientific information found in the media.
Improve critical thinking skills
Safely collect data and analyze data obtained through experiments.

^{*} Course Outcomes will be used in the Assessment Software for Outcomes Assessment. Limit to 3 - 5.